



# First synthesis of (–)-(1*R*,4*R*)- and (+)-(1*S*,4*S*)-(7,7-dimethyl-2-methylene-bicyclo[2.2.1]hept-1-yl)-methanol: new fenchone-derived chiral auxiliaries

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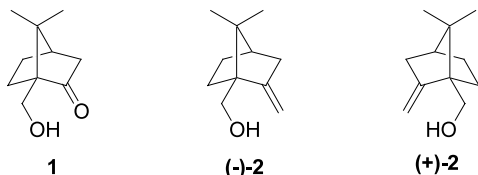
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**Abstract**—The treatment of (+)- or (–)-fenchones epoxides with 1:2 85% H<sub>3</sub>PO<sub>4</sub>:DMSO for 30 min at 20°C induced an enantiospecific Wagner–Meerwein rearrangement affording new C(10)-*O*-substituted camphor derivatives which could prove useful as chiral auxiliaries. © 2001 Elsevier Science Ltd. All rights reserved.

Bicyclic optically active derivatives of (*R*)-(+)-camphor have become of widespread use as chiral auxiliaries<sup>1</sup> as well as starting materials in asymmetric synthesis.<sup>2</sup> Among these, since the pioneering work of Oppolzer,<sup>1</sup> C(10)-substituted derivatives have been found to be of particular interest. Readily available 10-camphorsulfinic acids have been used as the starting material for the preparation of the majority of these derivatives which, consequently, have a sulphur atom attached to C(10).

García Martínez et al.<sup>3</sup> recently emphasized that other C(10)-heterosubstituted camphor derivatives have been prepared and tested as valuable chirality sources, among which C(10)-*O*-derivatives deserve special attention. These factors led these authors to finalise an efficient new preparation of 10-hydroxycamphor **1**.

These results prompted us to report herein our own findings in that field: a two step, enantiospecific preparation of alcohol **2**, which is another potentially interesting C(10)-*O*-substituted camphor derivative.

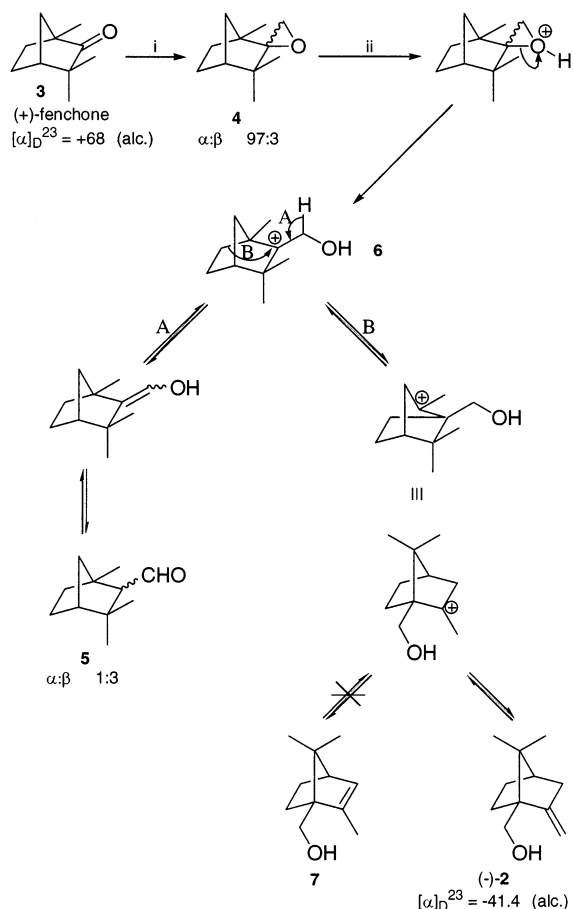


We previously reported<sup>4</sup> on the preparation of ethyl  $\alpha$ - and  $\beta$ -fenchols from fenchone **3**. Among the several routes we tried, one involved the (then unknown)  $\alpha$ - and  $\beta$ -fenchone epoxides **4**, which were thus prepared for the first time. However, we stated that ‘even stored at 0°C, unstable  $\alpha$ - and  $\beta$ -epoxides rearrange slowly into two compounds whose structures are now under investigation’.

Indeed, after heating overnight in refluxing chloroform (0.6 M solution) under argon, epoxides **4** ( $\alpha$ : $\beta$  97:3) were fully transformed (GC monitoring) into three new compounds which were easily separated by liquid chromatography on silica gel (cyclohexane:ethyl acetate, 96:4) and characterized as aldehyde **5**<sup>5</sup> ( $\alpha$ : $\beta$  1:3, 53%, not separated) and the as yet unreported alcohol **2**<sup>6</sup> (47%).

In order to explain the formation of **2** and **5**, the plausible route depicted in Scheme 1 can be considered. Traces of HCl dissolved in chloroform could protonate the epoxides **4** whose strain would then be released by opening into the tertiary carbocation **6**. The latter could next react following two pathways. Direct proton elimination (A) would afford the enol tautomer of aldehydes **5**. According to the other pathway (B), a Wagner–Meerwein rearrangement would lead to another tertiary carbocation whose stabilisation through proton elimination would then afford alcohol **2**. It is worth noting that, under these conditions, the isomeric alcohol **7** was not detected.

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**Scheme 1.** Reagents and conditions: (i) 3 equiv. of dimethylsulfonium methylide, DMSO/THF, -5°C, 1 h;<sup>4</sup> (ii) CHCl<sub>3</sub>, reflux, 21 h (Table 1, entry 1). Starting from (-)-fenchone of  $[\alpha]_D^{23} = -57.0$ , the same sequence afforded (+)-**2** of  $[\alpha]_D^{23} = +25.9$ .

As emphasized above, due to the interesting structural features of alcohol **2**, it was deemed useful to seek conditions that could favour its formation over that of aldehydes **5** (Table 1).

In this respect, Lewis acids (Table 1, entries 2–5) were ineffective as they mainly favoured the formation of aldehyde **5**, which is the sole product with SnCl<sub>4</sub> or Me<sub>2</sub>AlCl.

Turning to Brønsted acids proved more successful, especially when used in conjunction with a polar solvent. The best result was thus obtained after stirring for 30 min at room temperature in a 1:2 mixture of 85% H<sub>3</sub>PO<sub>4</sub> and DMSO (Table 1, entry 12). A small proportion (7%) of the isomeric alcohol **7** was formed in this case. Liquid chromatography on silica gel (cyclohexane:ethyl acetate, 96:4) allowed easy separation of aldehydes **5** but not of alcohol **7**. Alcohol **2** was therefore obtained in 65% yield from epoxides **4** as a 92:8 mixture with **7**.<sup>8</sup>

The enantiomeric purities of both (-)-**2** (derived from (+)-fenchone:<sup>9</sup>  $[\alpha]_D^{23} +68.0$  (alc.): o.p. = 0.9789) and (+)-**2** (from (-)-fenchone:  $[\alpha]_D^{23} -57.0$  (alc.): o.p. = 0.820) were determined by chiral GC analysis using a CP-Chirasil-DEX-CB capillary column (Chrompack, 25 m×0.25 mm i.d.). As shown in Fig. 1, the enantiomeric excesses measured for (-)-**2** (97.8% e.e.) and (+)-**2** (81.1% e.e.) were found to be identical to the enantiomeric purities of the starting (+)- and (-)-fenchones, respectively. Thus, the Wagner–Meerwein rearrangement occurred in an enantiospecific way.

In summary, starting from commercially available (+)- and (-)-fenchones we have developed a two-step enantiospecific synthesis of new C(10)-O-substituted chiral

**Table 1.** Acid-catalysed rearrangement of epoxides **4**<sup>a</sup>

Entry	Reagent	Solvent	Temp. (°C)	Time (h)	Ratio <b>2:5</b> :others <sup>b</sup>
1	HCl (traces)	Chloroform	Reflux	21	47:53
2	BF <sub>3</sub> ·OEt <sub>2</sub> (0.2 equiv.)	Diethyl ether	Rt	2.5	15:85
3	SnCl <sub>4</sub> (1 equiv.)	Dichloromethane	-60 to -20	2	0:100
4	Eu(Fod) <sub>3</sub> (0.1 equiv.)	Chloroform	Rt	2	48:52
5	Me <sub>2</sub> AlCl (0.5 equiv.)	Dichloromethane	-78	1	0:100
6	TsOH (0.05 equiv.)	Toluene	Rt	0.5	51:49
7	Amberlyst 15 (0.01 equiv.)	Hexane	Rt	18.5	8:92
8	3% HCl <sup>c</sup>	Methanol	0	0.25	56:22:22 <sup>d</sup>
9	HCl (gas)	Diethyl ether	Rt	0.50	62:38
10	85% H <sub>3</sub> PO <sub>4</sub> :diethyl ether (1:1)		Reflux	6	72:11:5:12 <sup>e</sup>
11	1 M H <sub>3</sub> PO <sub>4</sub> :dimethylsulfoxide (1:1)		Rt	0.25	73:10:7:10 <sup>f</sup>
12	85% H <sub>3</sub> PO <sub>4</sub> :dimethylsulfoxide (1:2)		Rt	0.5	82:11:7 <sup>g</sup>

<sup>a</sup> Reaction monitored by either TLC or GC. The indicated times refer to the disappearance of epoxides **4**.

<sup>b</sup> The product ratio was determined by both GC and 400 MHz <sup>1</sup>H NMR integration of the crude mixture.

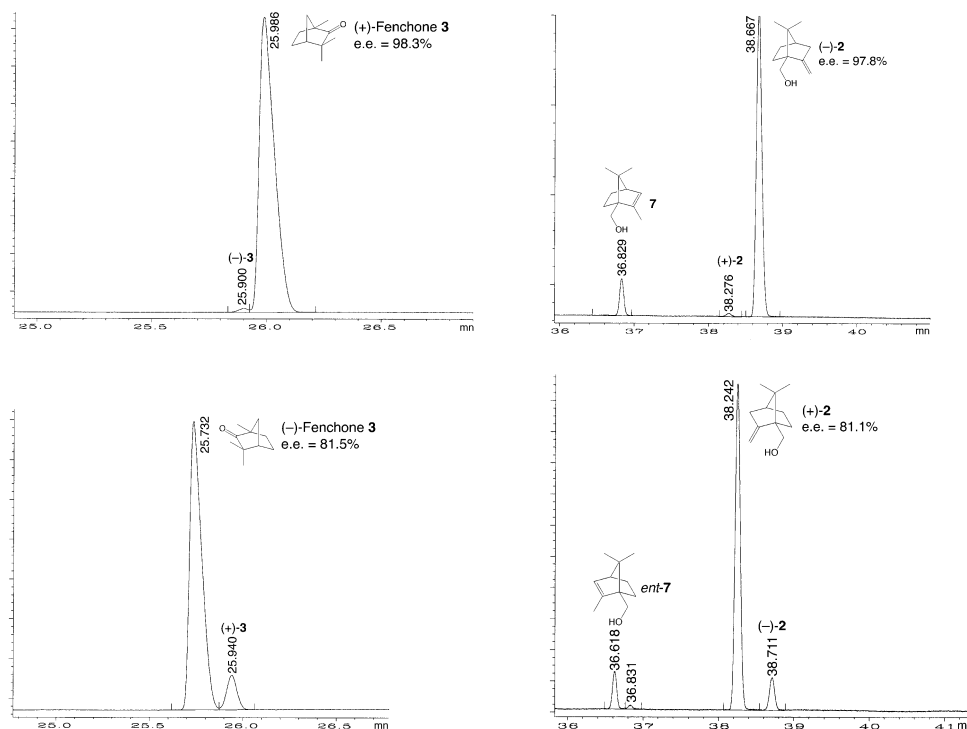
<sup>c</sup> Anhydrous HCl in methanol prepared by the addition of acetyl chloride (1 mL) to anhydrous methanol (20 mL).<sup>7</sup>

<sup>d</sup> The addition compound of methanol on the exocyclic double bond of **2** was detected.

<sup>e</sup> The isomeric alcohol **7** was observed (5%) together with the diol (12%) resulting from hydration of the double bond in either **2** or **7**.

<sup>f</sup> Others are alcohol **7** (7%) and diol (10%).

<sup>g</sup> The isomeric alcohol **7** (7%) was detected.



**Figure 1.** GLC analysis on a CP-Chirasil-DEX-CB capillary column (Chrompack, 25 m×0.25 mm i.d.) of the starting (+)- and (-)-fenchones and of the alcohols resulting from the treatment of epoxides **4** with 85%  $\text{H}_3\text{PO}_4$  in DMSO (Table 1, entry 12). In both cases, the e.e. of the resulting alcohols **2** and **7** are found identical to the e.e. of the starting fenchones.

camphor derivatives. The possibility of the development of these derivatives as chiral auxiliaries is currently under investigation.

## References

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- The structure of **2** was deduced from the following spectroscopic data and from C-H COSY and HMBC NMR experiments. (-)-**2**: white solid, mp 75.5–78.4°C (sublim.);  $[\alpha]_{\text{D}}^{23}$  -41.4 (c 0.9, alc.);  $\nu_{\text{max}}$  (KBr)/ $\text{cm}^{-1}$  3394, 2938, 2875, 1654, 1450, 1386, 1370, 1020, 873;  $\delta_{\text{H}}$  (400 MHz,  $\text{CDCl}_3$ ) 0.95 (s, 3H), 1.00 (s, 3H), 1.26 (m, 2H), 1.56 (br. s, 1H), 1.70 (br. t,  $J=4.3$  Hz, 1H), 1.72–1.88 (m, 2H), 1.94 (br. d,  $J=16.3$  Hz, 1H), 2.47 (br. d,  $J=16.3$  Hz, 1H), 3.82 (AB,  $J_{\text{AB}}=11.7$  Hz, 2H), 4.80 (br. t,  $J=2.0$  Hz, 1H), 4.85 (br. t,  $J=2.0$  Hz, 1H);  $\delta_{\text{C}}$  (100 MHz,  $\text{CDCl}_3$ ) 19.47 (q), 21.09 (q), 27.63 (t), 31.18 (t), 37.19 (t), 45.80 (d), 47.54 (s), 56.27 (s), 62.26 (t), 102.59 (t), 156.08 (s);  $m/z$  (GC/MS, EI 70 eV, % rel.) 166(8), 148(28), 135(81), 123(59), 105(82), 93(100), 81(77), 67(60), 55(33), 41(80);  $m/z$  (GC/MS, CI isobutane, % rel.) 167(27), 149(100), 135(2), 121(16), 107(23).
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- Preparative details (Table 1, entry 12): a room temperature cooled mixture of DMSO (40 mL) and 85%  $\text{H}_3\text{PO}_4$  (20 mL) was added to epoxide **4** (2.496 g, 15 mmol). After stirring for 30 min at rt, GC analysis of an aliquot showed the complete disappearance of epoxide **4**. 50% aq. NaOH (60 mL) solution was then added. The resulting white suspension was diluted with water (200 mL), filtered and the filtrate extracted with pentane (3×100 mL). The residue was dissolved in water (150 mL) and extracted with pentane (3×50 mL). The combined organic layers were washed with brine (2×100 mL), dried over  $\text{MgSO}_4$  and the solvent was removed in vacuo to afford a white solid (2.043 g). Liquid chromatography on silica gel (cyclohexane:ethyl acetate, 96:4) gave two fractions: aldehydes **5** (0.200 g, 8% yield) and unseparated isomeric alcohols **2** and **7** (1.622 g, 65% combined yield, ratio of **2**:**7**=92:8).
- (+)-Fenchone:  $[\alpha]_{\text{D}}^{16}=+69.5$  (alc.); Günther, E. *The essential Oils*; Van Nostrand: New York, 1948; Vol. II, p. 420.